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Organic chemists use a curved arrow ( $\curvearrowright$ ) to show the movement of an electron pair. A sigma ( $\sigma$ ) bond is formed between atoms by the overlap of two atomic orbitals along the line that connects the atoms. Carbon uses  $sp^3$ -hybridized orbitals to form four such bonds. These bonds are directed from the carbon nucleus toward the corners of a tetrahedron.

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## BASIC PRINCIPLES OF ORGANIC CHEMISTRY

The goal is not to understand every single thing about this organic chemistry mechanism in great detail right now. Our goal right now is just to appreciate how nucleophiles and electrophiles are used in organic chemistry mechanisms and start to get a feeling for how these curved arrows show the movement or flow of electrons.

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Intro to organic mechanisms (video) | Khan Academy

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